



Surname \_\_\_\_\_

Forename(s) \_\_\_\_\_

Centre Number \_\_\_\_\_

Candidate Number \_\_\_\_\_

Candidate Signature \_\_\_\_\_

I declare this is my own work.

**A-level**

**PHYSICS**

**Paper 3**

**Section B Turning points in physics**

**7408/3BD**

**Monday 17 June 2024**

**Morning**

**Time allowed: The total time for both sections of this paper is 2 hours. You are advised to spend approximately 50 minutes on this section.**

**[Turn over]**



On the front of this book, write your surname and forename(s), your centre number, your candidate number and add your signature.

## **MATERIALS**

For this paper you must have:

- a pencil and a ruler
- a scientific calculator
- a Data and Formulae Booklet
- a protractor.

## **INSTRUCTIONS**

- Use black ink or black ball-point pen.
- Answer ALL questions.
- You must answer the questions in the spaces provided. Do not write outside the box around each page or on blank pages.
- If you need extra space for your answer(s), use the lined pages at the end of this book. Write the question number against your answer(s).
- Do all rough work in this book. Cross through any work you do not want to be marked.
- Show all your working.



## **INFORMATION**

- **The marks for questions are shown in brackets.**
- **The maximum mark for this paper is 35.**
- **You are expected to use a scientific calculator where appropriate.**
- **A Data and Formulae Booklet is provided as a loose insert.**

**DO NOT TURN OVER UNTIL TOLD TO DO SO**







0	1	.	2
---	---	---	---

Determine, using the following data, a value for the specific charge of the electron.

$$B = 1.59 \text{ mT}$$

$$V_P = 1.51 \text{ kV}$$

$$d = 5.0 \text{ cm}$$

$$V_A = 1.00 \text{ kV}$$

[4 marks]

specific charge = \_\_\_\_\_  $\text{C kg}^{-1}$

6



**BLANK PAGE**

**[Turn over]**

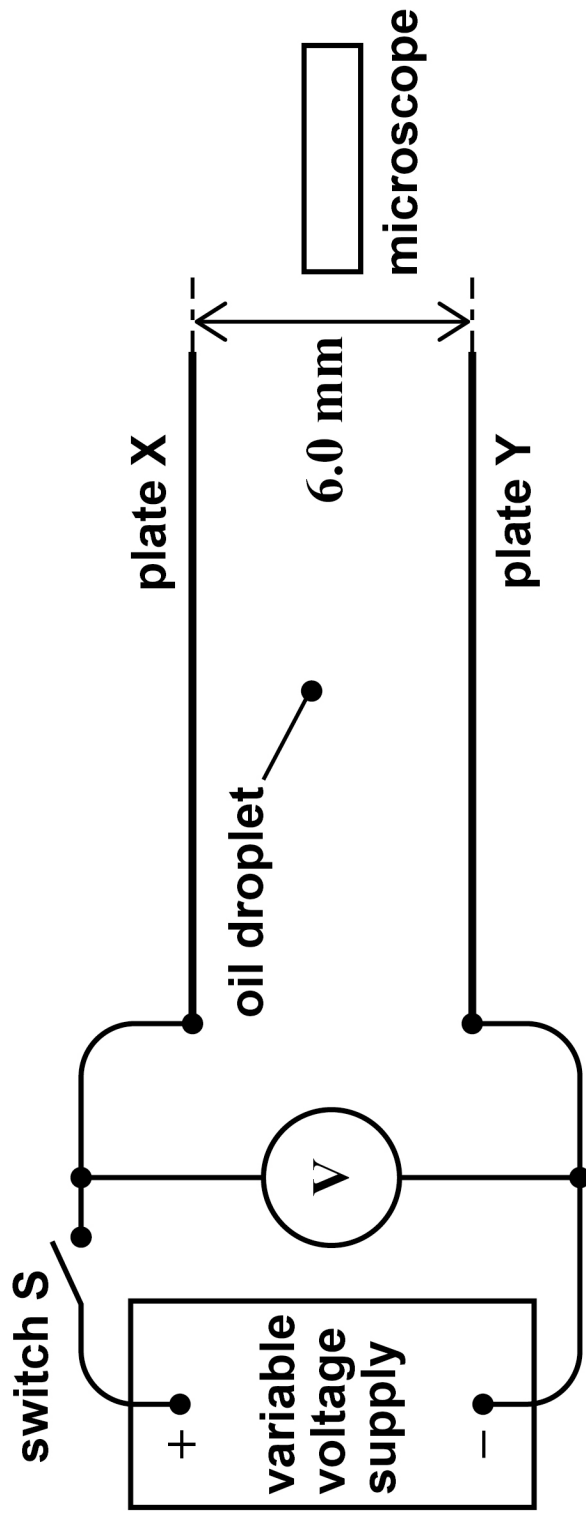




0 2 . 1

FIGURE 2 shows a cross-sectional view of the arrangement that Millikan used to determine the charge on the electron.

FIGURE 2





**Millikan's initial step was to determine the radius of the oil droplet.**

**Explain how Millikan used this apparatus to determine the radius of the oil droplet.**

**In your answer you should:**

- describe the procedure used, the measurements taken and any additional data required**
- describe how the radius was determined from the measurements**
- state the physical principles and assumptions involved in the determination of the radius.**

**[6 marks]**

---

---

---

**[Turn over]**

A series of 10 vertical lines spaced evenly across the page, likely serving as a guide for writing or a decorative element.



---

---

---

---

---

---

---

---

---

---

**[Turn over]**



--	--	--	--	--	--	--	--	--	--	--



---

---

---

---

---

---

---

**[Turn over]**



0	2	.	2
---	---	---	---

On one occasion, the radius of a droplet was determined to be  $1.20 \times 10^{-6}$  m.

When the droplet was stationary, the voltmeter reading was 467 V.

Show that the charge on the droplet was approximately  $8 \times 10^{-19}$  C.

density of oil =  $880 \text{ kg m}^{-3}$

[3 marks]



**BLANK PAGE**

**[Turn over]**



**02.3**

**TABLE 1** shows the percentage uncertainty in each quantity.

**TABLE 1**

<b>QUANTITY</b>	<b>PERCENTAGE UNCERTAINTY</b>
<b>radius of oil droplet</b>	<b>4%</b>
<b>density of oil</b>	<b>1%</b>
<b>gravitational field strength</b>	<b>0.1%</b>
<b>potential difference</b>	<b>0.2%</b>
<b>distance between the plates</b>	<b>2%</b>



Show that the absolute uncertainty in your answer to Question 02.2, on page 14, is approximately  $\pm 1 \times 10^{-19}$  C.

Go on to discuss whether this uncertainty allows your answer to Question 02.2 to be used to support the quantisation of electric charge. [3 marks]

---

---

---

---

---

---

---

---

[Turn over]



---

---

---

---

---

---

---

---

---

---

12



**BLANK PAGE**

**[Turn over]**



0	3
---	---

**Hertz did an experiment to determine the speed of radio waves.**

**Describe this experiment.**

**In your answer you should:**

- **include a labelled diagram**
- **state the measurements that were taken**
- **describe how the data were used to determine the speed of radio waves.**

**[5 marks]**



---

---

---

---

---

---

---

---

---

**[Turn over]**





---

---

---

---

[Turn over]

5

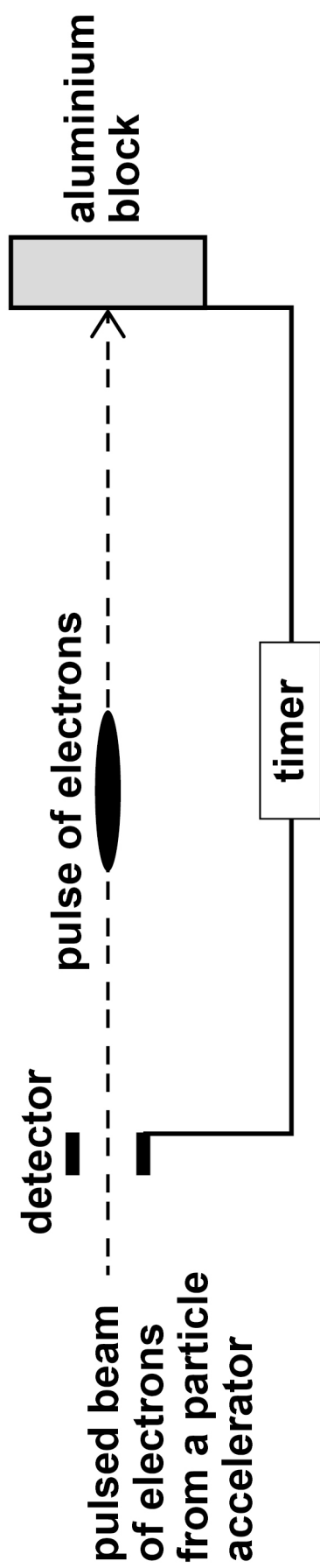




0 4

**FIGURE 3** shows a modern version of Bertozzi's experiment to measure the kinetic energy of high-speed electrons. A timer is used to measure the time taken for a pulse of electrons to travel from the detector to the aluminium block.

**FIGURE 3**





0 4 . 1

A potential difference (pd) of 1.30 MV is used to accelerate the electrons.

Show that each electron gains approximately  $2 \times 10^{-13}$  J of kinetic energy.

[1 mark]

[Turn over]



0 4 . 2

These electrons cause the temperature of the aluminium block to increase by 68.0 K.

The number of electrons that cause this increase in temperature is  $4.50 \times 10^{17}$

Deduce whether this increase in temperature is consistent with an accelerating pd of 1.30 MV.

specific heat capacity of aluminium =  $903 \text{ J kg}^{-1} \text{ K}^{-1}$

mass of aluminium block = 1.50 kg

[2 marks]



---

---

**[Turn over]**



0 4 . 3

The speed of the electrons between the detector and the block is  $2.88 \times 10^8 \text{ m s}^{-1}$ .

Student A suggests that the non-relativistic equation for kinetic energy could be used.

Student B suggests that the relativistic equation for kinetic energy is required in this situation.

Evaluate the suggestions of student A and student B.  
Support your answer with calculations. [4 marks]

---

---

---

**[Turn over]**



Five vertical lines for writing.





0 4 . 4

The timer in FIGURE 3, on page 24, records a time of 29.8 ns.

What is the proper time interval for an electron travelling from the detector to the aluminium block? [1 mark]

Tick (✓) ONE box.

< 29.8 ns

29.8 ns

> 29.8 ns

[Turn over]



04.5

The electrons in FIGURE 3, on page 24, were accelerated from rest in 13 stages.

In each stage the electrons were accelerated by a pd of 100 kV.

As a result, an electron increases its speed and kinetic energy during each stage.

Compare, for an electron,

- its increase in speed for the first stage with that for the last stage
- its increase in kinetic energy for the first stage with that for the last stage.

32

Justify your answer.

No further calculations are required. [4 marks]

---

---

---



---

---

---

---

---

---

**END OF QUESTIONS**





**BLANK PAGE**

For Examiner's Use	
Question	Mark
1	
2	
3	
4	
<b>TOTAL</b>	

**Copyright information**

For confidentiality purposes, all acknowledgements of third-party copyright material are published in a separate booklet. This booklet is published after each live examination series and is available for free download from [www.aqa.org.uk](http://www.aqa.org.uk).

Permission to reproduce all copyright material has been applied for. In some cases, efforts to contact copyright-holders may have been unsuccessful and AQA will be happy to rectify any omissions of acknowledgements. If you have any queries please contact the Copyright Team.

Copyright © 2024 AQA and its licensors. All rights reserved.

**WP/M/DL/Jun24/7408/3BD/V3**